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Chemistry IF8766 9 Olnstructional Fair, Inc. 59. CALCULATIONS USING SIGNIFICANT FIGURES Name? When multiplying and dividing, limit and round to the least number of significant figures in any of the factors. 11 Example 1: 23.0 cm x 432 cm x 19 cm = 188,784 cm3 11

SIGNIFICANT FIGURES Name

Naming Ionic Compounds Directions: Name the following compounds using the Stock Naming System. 1. CaCO 3 ... Chemistry IF8766 Page 45 Instructional Fair, Inc. Title: Name: Author: ASD Created Date: 1/17/2010 4:03:36 PM ...

Naming Ionic Compounds - Ms. Agostine's Chemistry Page

IDEAL GAS LAW Use the ideal Gas Law below to solve the following problems. pressure in atmospheres volume in liters number of moles L atm Universal Gas Constant = 0.0821

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enstructional Fair, Inc. 5. 35 g of CuSOa,5H20 Determine the number of grams in each of the quantities below. 1, 2.5 moles of NaCl 2. 0.50 moles of 3, 1.70 moles of KMn04 4, 0.25 moles of KCI e, ST 5. 3.2 moles of CuSO405H20 Chemistry IF8766

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r i c Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name _ Ionicbondingoccurswhenametcl hcnstersoneormoreelectrons10anonmelolln on ...

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Writing Formulas From Names Directions: Write the formulas of the following compounds. 1. ammonium phosphate (NH 4) 3 PO ... Chemistry IF8766 Page 48 Instructional Fair, Inc. Title: Name: Author: ASD

Writing Formulas From Names - Ms. Agostine's Chemistry Page

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NAMING HYDROCARBONS Name Name the compounds below according to the IUPAC naming system 1. H H H I I I H — C — C — C — H I I I H H H 5.

NAMING HYDROCARBONS Name

©Instructional Fair, Inc. 20 Chemistry IF8766. CHARLES' LAW Name charles' Law states that the volume of a gas varies directly with the Kelvin temperature, assuming that pressure is constant. We use the following formulas:

- or x = x T K oc + 273

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Boyle's Law - Solutions 1) If I have 5.6 liters of gas in a piston at a pressure of 1.5 atm and compress the gas until its volume is 4.8 L, what will the new pressure inside the piston be?

Boyles Law Worksheet with anwer key

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